



ISSN 2250-2688  
Received: 15/03/2012  
Revised: 24/03/2012  
Accepted: 29/03/2012

**Padmini Shukla, Prabodh Shukla**  
Pranveer Singh Institute of Technology,  
Kanpur, (U.P.)

**B. Gopalakrishna**  
R.R. College of Pharmacy,  
Chikbanawara, Bangalore

## Isolation and Characterization of $\beta$ -sitosterol from *Cissampelos pareira* Linn.

**Padmini Shukla \*, Prabodh Shukla, B. Gopalakrishna**

### ABSTRACT

General phytochemical screening of *Cissampelos pareira* (Menispermaceae) revealed the presence of sterols, polyphenolic compounds, saponins, alkaloids etc. The aim of this study is to identify and characterize the bioactive principle from the plant. It has wide folk medicinal uses. The isolation and characterization of Phytoconstituent was done from the chloroform extract of *Cissampelos pareira* Linn. Chloroform extract was subjected to column chromatography and eluted with solvent mixtures of petroleum ether, benzene and chloroform to isolate phytoconstituent. The structure of the isolated compound was established on the basis of elemental analysis and spectroscopic evidences (IR, UV,  $^1\text{H}$ NMR,  $^{13}\text{C}$ NMR, MS). A sterol, stigmast-5-en-3 $\beta$ -ol was isolated from the chloroform extract of the plant. The yield of the compound was 0.0042% w/w, m.p. 136- 138 $^{\circ}\text{C}$ , Rf value 0.7 in Toluene: diethylether: Cyclohexane (6:3:1). *Cissampelos pareira* contains  $\beta$ -sitosterol which may be responsible for various pharmacological activities of the plant.

**Keywords:** *Cissampelos pareira*,  $\beta$ -sitosterol, Menispermaceae, Chloroform extract.

### 1. INTRODUCTION

*Cissampelos pareira* of the family Menispermaceae are commonly known as Patha in Ayurveda and have been used for the treatment of fever, urinary problems and skin infections.<sup>1</sup> *Cissampelos pareira* is found very common in semi dry forests of tropics.<sup>2</sup> Various alkaloids and different pharmacological activities of these plants have been reported. Hydrocolloids,<sup>3</sup> Cissampeloflavone,<sup>4</sup> A tropone-isoquinoline alkaloid, pareitropone,<sup>5</sup> Bisbenzylisoquinoline alkaloids, cissamperine with tumor inhibitor activity,<sup>6</sup> trandrone,<sup>7</sup> tropoloisoquinoline alkaloids such as pareirubrine A and B with antileukemic activity<sup>8</sup> have been isolated from *C. pareira* Var. *hirsuta*. Plant extracts were tested for Antipyretic activity,<sup>9</sup> Chemomodulatory influence of Hirsuta on Gastric cancer and antioxidant system in experimental animal,<sup>10</sup> anti-inflammatory activity,<sup>9</sup> Immunomodulatory activity,<sup>10</sup> antifertility activity, antinociceptive, antiarthritic activity, antibacterial activity,<sup>11</sup> antimalarial activity,<sup>12</sup> diuretic activity,<sup>13</sup> hypoglycemic activity<sup>14</sup> and anticonvulsant activity.<sup>15</sup> A few ethnobotanical reports on treatment of fever,<sup>16</sup> gastrointestinal tract disorders<sup>17</sup> were also investigated. The plant also contains some other phytoconstituents like sterols, saponins, flavonols etc along with alkaloid.<sup>18,19</sup> The present study is an attempt to isolate and characterize  $\beta$ - sitosterol from the aerial parts of *Cissampelos pareira* Linn.

**Correspondence:**  
**Padmini Shukla**  
Pranveer Singh Institute of Technology,  
Kanpur (U.P.)  
E-mail: pdmmishra@gmail.com

## 2. MATERIALS AND METHODS

### Collection & Authentication of plant Material

The plants of *Cissampelos pareira* Linn were collected from the National botanical research institute, Lucknow, Uttar Pradesh, India and were authenticated by Prof J. P. Shukla, Department of Botany, D.B.S College, Kanpur where a voucher specimen (PH/CP/11) is deposited for further reference.

### Extraction and isolation of compound

The shade dried and coarsely powdered leaves were successively extracted with increasing polarity solvents like petroleum ether, chloroform and alcohol using soxhlet apparatus. Chloroform extract was packed in a column containing silica gel 'H' and eluted with solvent mixtures of petroleum ether, benzene and chloroform. All the fractions were monitored on TLC. The fractions collected with petroleum ether: benzene (20:80) was pooled together. The TLC of these fraction was performed by using solvent system Toluene: diethylether: Cyclohexane (6:3:1) which shows a spot at 0.70 Rf value. It was evaporated on a water bath (50- 60°C) to afford a solid residue. The residue was dissolved in a mixture of Chloroform:Ethanol (40:60) with little warming on a water bath. It was left undisturbed in refrigerator when needle shaped crystals was obtained. The characterisation of the isolated compound was done on the basis of phytochemical analysis (Salkowski reaction, Liebermann burchard reaction) and spectroscopic studies. (IR, UV, <sup>1</sup>HNMR, <sup>13</sup>C-NMR & FAB-MS).

### Tests for alcohol

4g of ceric ammonium nitrate was dissolved in 10ml of 2N HNO<sub>3</sub> on mild heating. A few crystals of isolated compound were dissolved in 0.5ml of dioxane. The solution was added to 0.5ml of ceric ammonium nitrate reagent and diluted to 1ml with dioxane and shaken well. The developed yellow to red color indicates the presence of an alcoholic hydroxyl group.<sup>20</sup>

### Tests for steroid

#### Salkowski reaction

A few crystals were dissolved in chloroform and a few drops of concentrated sulfuric acid were added to the solution. A reddish color was seen in the upper chloroform layer.<sup>20</sup>

#### Liebermann burchard reaction

A few crystals were dissolved in chloroform and a few drops of concentrated sulfuric acid were added to it followed by addition of 2-3 drops of acetic anhydride. Solution turned violet blue and finally green.<sup>20</sup>

## 3. RESULTS AND DISCUSSION

Phytochemical analysis (Salkowski's test and Liebermann-Burchard test) of the compound confirms its steroidal nature. The elemental analysis (Elementar, Vario EL III) revealed that the compound contains 83.86% of C, 12.25% of H and 3.89% of O. The N % was found to be nil. Now based on the number of O (1 or 2) in the proposed compound the molecular weight could be: 411.30 or 822.62 respectively. So, Based upon the number of O the formula could be C<sub>29</sub>H<sub>50</sub>O or C<sub>58</sub>H<sub>100</sub>O<sub>2</sub>. From <sup>13</sup>CNMR and <sup>1</sup>HNMR the number of C and H was found to be near to the first formula i.e. C<sub>29</sub>H<sub>50</sub>O. The exact molecular mass for the formula was found to be 414.7.

The IR absorption spectrum showed absorption peaks at 3373.6cm<sup>-1</sup> (O-H stretching.); 2940.7 cm<sup>-1</sup> and 2867.9 cm<sup>-1</sup> (aliphatic C-H stretching); 1641.6 cm<sup>-1</sup> (C=C absorption peak); other absorption peaks includes 1457.3 cm<sup>-1</sup> (CH<sub>2</sub>); 1381.6 cm<sup>-1</sup> (OH def), 1038.7 cm<sup>-1</sup> (cycloalkane) and 881.6 cm<sup>-1</sup>.

<sup>1</sup>HNMR (CDCl<sub>3</sub>, 400MHz): <sup>1</sup>HNMR has given signals at δ 3.2(1H, m, H-3), 5.26 (1H, m, H-6), 5.19(1H, m, H-23), 4.68(1H, m, H- 22), 3.638(1H, m, H-3), 2.38(1H, m, H-20), 1.8-2.0 (5H, m) ppm. Other peaks are observed at δ 0.76-0.89 (m, 9H), 0.91-1.05 (m, 5H), 1.35-1.42 (m, 4H), 0.69-0.73 (m, 3H), 1.8-2.00 (m, 5H), 1.07-1.13 (m, 3H), 1.35-1.6 (m, 9H) ppm.

<sup>13</sup>CNMR (CDCl<sub>3</sub>, 100MHz): <sup>13</sup>CNMR has given signal at 150.98, 145.2 (C-5), 139.8 (C-22), 121.7, 118.89(C-6), 79.03 (C-3), 55.3(C-14), 55.18(C-17), 50.45 (C-9), 48.3 (C-9), 40.8 (C-20), 40.1(C-12), 39.2 (C-13), 38.9 (C-4), 38.6 (C-12), 37.18 (C-1), 37.12 (C-10), 36.3 (C-8), 35.59(C-20), 34.29 (C-22), 34.24 (C-7), 32.66 (C-8), 29.86 (C-25), 29.71 (C-16), 28.41 (C-2), 28.1 (C-15), 27.4 (C-28), 26.1 (C-11, 26), 21.6 (C-27), 19.32 (C-19), 17.71 (C-21), 15.6 (C-18, 29).

FAB-MS spectroscopy showed the molecular ion peaks at 414 that correspond to molecular formula, C<sub>29</sub>H<sub>50</sub>O. Ion peaks were also observed at m/z 367, 271, 255, 229, 189, 175, 161, 133, 121, 105, 107, 95, 81, 69, 55, 41.

Based on the melting point and other related data (IR, NMR and Mass) the structure of the isolated compound was proposed as

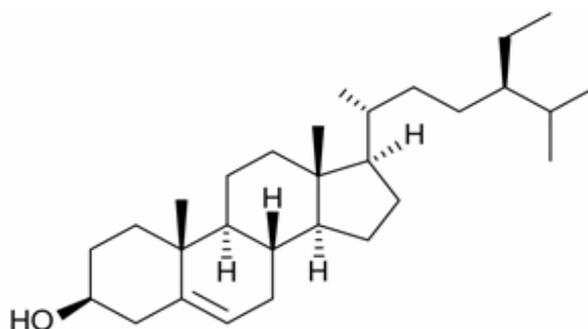


Figure 1: β-sitosterol (C<sub>29</sub>H<sub>50</sub>O; Mol.Wt: 414.71)

#### 4. CONCLUSION

From the above study,  $\beta$ -Sitosterol was isolated and characterized from chloroform extract of *Cissampelos pareira* Linn. and this is a phytosterol.  $\beta$ -Sitosterol reduce carcinogen induced cancer of the colon. It shows antiinflammatory, anti-pyretic, antiarthritic, anti-ulcer, insulin releasing and oestrogenic effects and inhibition of spermatogenesis. Beta-sitosterol is mainly known and used for its cholesterol lowering property. But studies have shown that the phytochemical may have other health benefits: easing symptoms of benign prostatic enlargement, reducing risk of cancer and prevention of oxidative damage through its antioxidant activity.

#### REFERENCES

1. Yoganarasimhan SN *Medicinal Plants of India*, New Delhi: Interline publishing pvt. Ltd; 2002.
2. Saldanha CJ. *Flora of Karnataka*, New Delhi: Oxford and IBH publishing co; 1984.
3. Ramírez I, Carabot A, Meléndez P, Carmona J, Jimenez M, Patel AV, Crabb TA. Cissampeloflavone, a chalcone-flavone dimer from *Cissampelos pareira*, *Phytochemistry*. 2003; 64:645-47.
4. Fournet A, Cavé A, Duté P, Weber JF and Bruneton J. Bisbenzylisoquinoline alkaloids from *Abuta pahni*. *Phytochemistry*. 1987; 26: 2136-2137.
5. Bhakuni DS, Jain S, Chaturvedi R. The biosynthesis of the alkaloids of *cissampelos pareira* linn. *Tetrahedron*. 1987; 3975-3982.
6. Kupchan SM, Patel AC, Fujita E. Tumor inhibitors VI. Cissamperine, new cytotoxic alkaloid from *Cissampelos pareira* Var. *hirsuta*. Cytotoxicity of Bisbenzylisoquinoline alkaloids. *J. Pharm. Sci*. 1965; 4: 580-83.
7. Rojanasonthorn G. The isolation and characterization of bisbenzylisoquinoline alkaloid "tetrandrine" from the root of *Cissampelos pareira* Var. *hirsuta* l. [dissertation]. Bangkok, Thailand: Mahidol University; 1970.
8. Morita H, Matsumoto K, Takeya K, Itokawa H, Iitaka Y. Structures and solid state tautomeric forms of two novel antileukemic tropoloisoquinoline alkaloids, pareirubrines A and B, from *Cissampelos pareira*. *Chem Pharm Bull (Tokyo)*. 1993; 8:1418-22.
9. Morris GA, Castile J, Smith A, Adams GG Harding SE. The effect of prolonged storage at different temperatures on the particle size distribution of tripolyphosphate (TPP) – chitosan nanoparticles. *Carbohydrate polymers*. 2004:399-400.
10. Vardhanabhuti B, alkeda S. Isolation and characterization of hydrocolloids from monoi (*Cissampelos pareira*) leaves. *Food hydrocolloids*. 2006: 885-891.
11. Perez C, Anesini C. *In vitro* antibacterial activity of Argentine folk medicinal plants against *Salmonella typhi*. *J Ethnopharmacol*. 1994; 1: 41-46.
12. Gessler MC , Nkunyak MH , Mwasumbi LB, Heinrich M, Tanner M. Screening Tanzanian medicinal plants for antimalarial activity. *Acta Trop*. 1994:65-77.
13. Caceres A, Giron LM, Martinez AM. Diuretic activity of plants used for the treatment of urinary ailments in Guatemala. *J Ethnopharmacol*. 1987;19(3):233-45.
14. Tripathi SN, Tiwari CM, Upadhyay BN, Singh RS. Screening of hypoglycemic action in certain indigenous drugs. *J Res Indian Med Yoga Homeopathy*. 1979;14(3):159-69.
15. Adesina SK. Studies on some plants used as anticonvulsants in Amerindian and African traditional medicine. *Fitoterapia*. 1982; (53):147- 62.
16. Singh KK, Maheshwari JK. Traditional phytotherapy of some medicinal plants used by the tharus of the Nainital district, UttarPradesh, India. *Int J Pharmacog*. 1994; 32(1):51-58.
17. Caceres A, Cano O, Samayoa B, Aguilar L. Plants used in Guatemala for the treatment of gastrointestinal disorders. Screening of 84 plants against enterobacteria. *J Ethnopharmacol*. 1990; 30(1):55-73.
18. Gupta A, Pandey S, Shah DR, Seth NR, Yadav JS. Pharmacognostical and Phytochemical Evaluation of Leaves of *Cissampelos pareira* *Pharmacognosy Journal*. 2011; 21(3):25-28.
19. Anonymous The Ayurvedic Pharmacopoeia of India, Part 1, Vol.1. 1st ed. New Delhi: Govt of India, Ministry of Health and Family Welfare, Dept of Health; (1978) 92.
20. Harborne JB. *Phytochemical Methods: A Guide to Modern Techniques of Plant Analysis*. 3rd Edn., chapman and Hall: London. ISBN: 0-412-57270-2, (1998) 302.